

# Development of Primary Certified Reference Materials (PCRMTM)

Primary SI Traceable Solution  
Standards For Iridium and Osmium

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# What is a Primary Solution Standard?

**Measured value** is obtained by means of a **primary reference measurement procedure**.<sup>1, 2</sup>

**Gravimetry, or measurement by weight,** is a primary measurement procedure.

1. VIM\_JCGM\_200\_2012, section 5.4
2. VIM\_JCGM\_200\_2012, section 2.8

# What is SI Traceability?

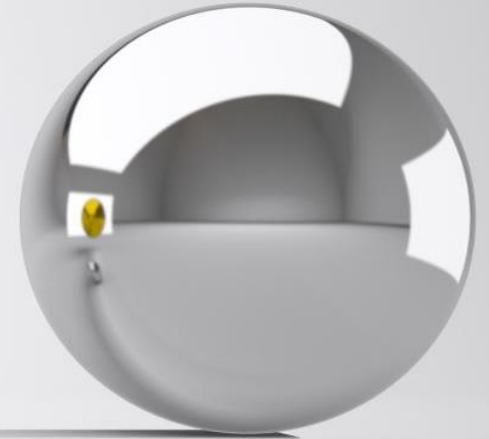
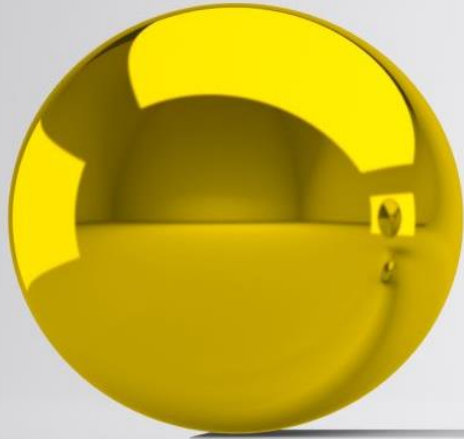
Your measured value is linked or traced to the International System of Units, the SI.

The SI was established in 1960.

There are 7 base units in the SI.<sup>3</sup>

| Base Quantity             | Base Unit | Symbol |
|---------------------------|-----------|--------|
| Time                      | Second    | s      |
| Length                    | Meter     | m      |
| Mass                      | Kilogram  | kg     |
| Electric Current          | Ampere    | A      |
| Thermodynamic Temperature | Kelvin    | K      |
| Amount of Substance       | Mole      | Mol    |
| Luminous Intensity        | Candela   | cd     |





## Establish SI traceability?

Our **primary measurement procedure** was **Gravimetry**, or measurement by weight.

The kilogram (kg) is one of the 7 base units of the SI.

Thus, we obtain traceability to the SI through the kg.

# Why develop a primary SI traceable solution standard?

Ensure the accuracy of our Os and Ir Certified Reference Materials (CRMs).

No National Metrology Institute (NMI) has an SI traceable solution standards for Os and Ir.



# Why develop a primary SI traceable solution standard?

- ISO is the International Organization for Standardization.
- Accreditation to the ISO 17034 and ISO 17025 standards is essential for a CRM manufacturer .
- Primary Requirement: Metrological traceability to the SI.





# Requirements for SI traceability?

Primary  
Measurement  
Procedure

Purity Analysis of  
the candidate  
Starting Materials

# Starting Materials

## Candidate SI traceable starting materials

- Os and Ir salts:
  - Ammonium hexachloroosmate
  - Ammonium hexachloroiridate hydrate

## High purity Os and Ir metal powders



# Experimental Design

Followed the approach used by Beck, Salit et al. at NIST when they developed the Rh SRM.<sup>1</sup>

Determine the Os or Ir mass fraction in the candidate metal salt.

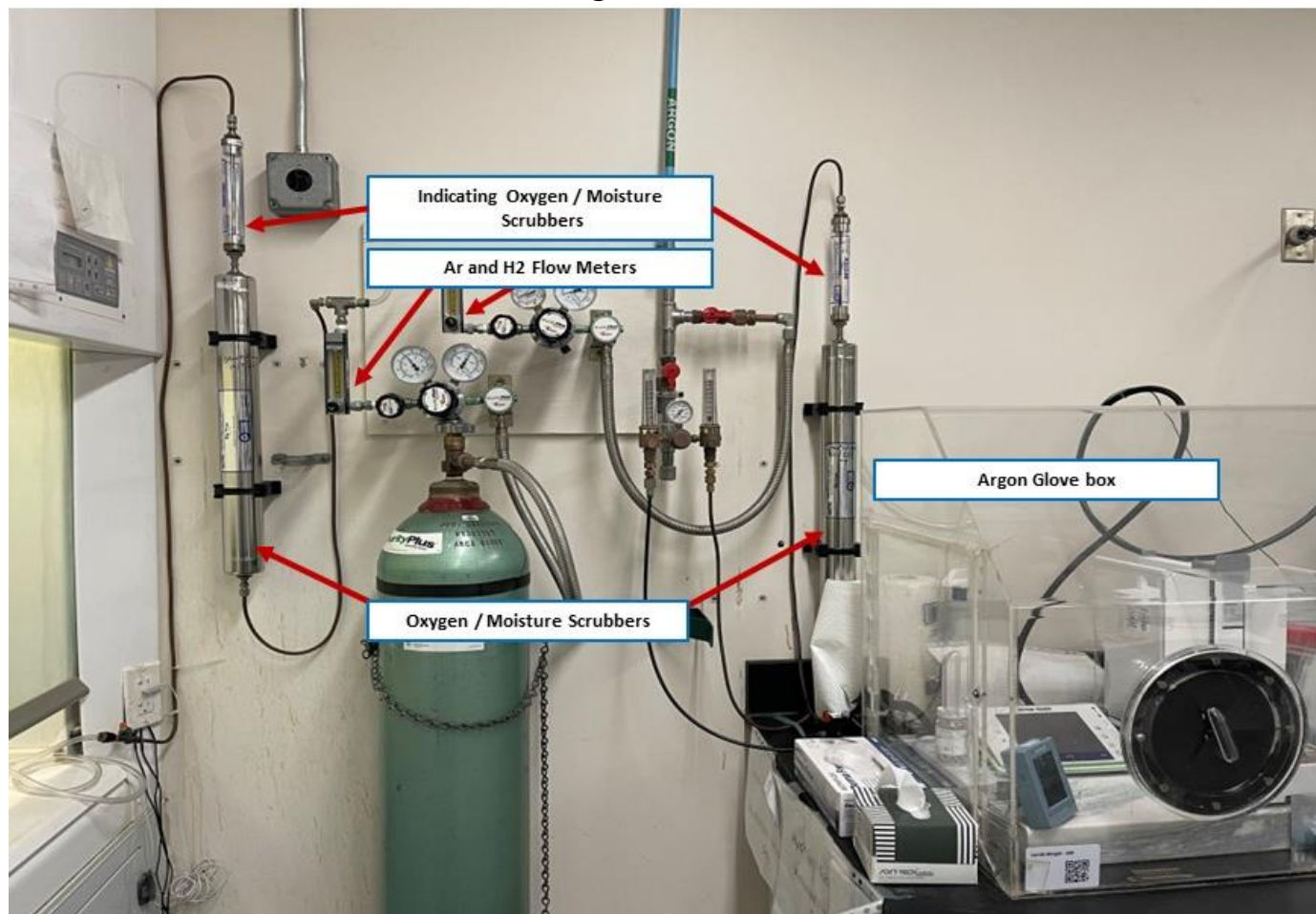
- Gravimetric Reductions of the salt to the metal
- Establishes SI traceability of the candidate starting material

High purity Os or Ir metal powder

- An independent source of Os or Ir
- Used as a standard for comparison

<sup>1</sup>Beck, C. M., II; Salit, M. L. Preparation and Certification of a Rhodium Standard Reference Material Solution. Anal. Chem. 1993, 65, 2899– 2902, <https://doi.org/10.1021/ac00068a030>

# Eliminate Systematic Errors

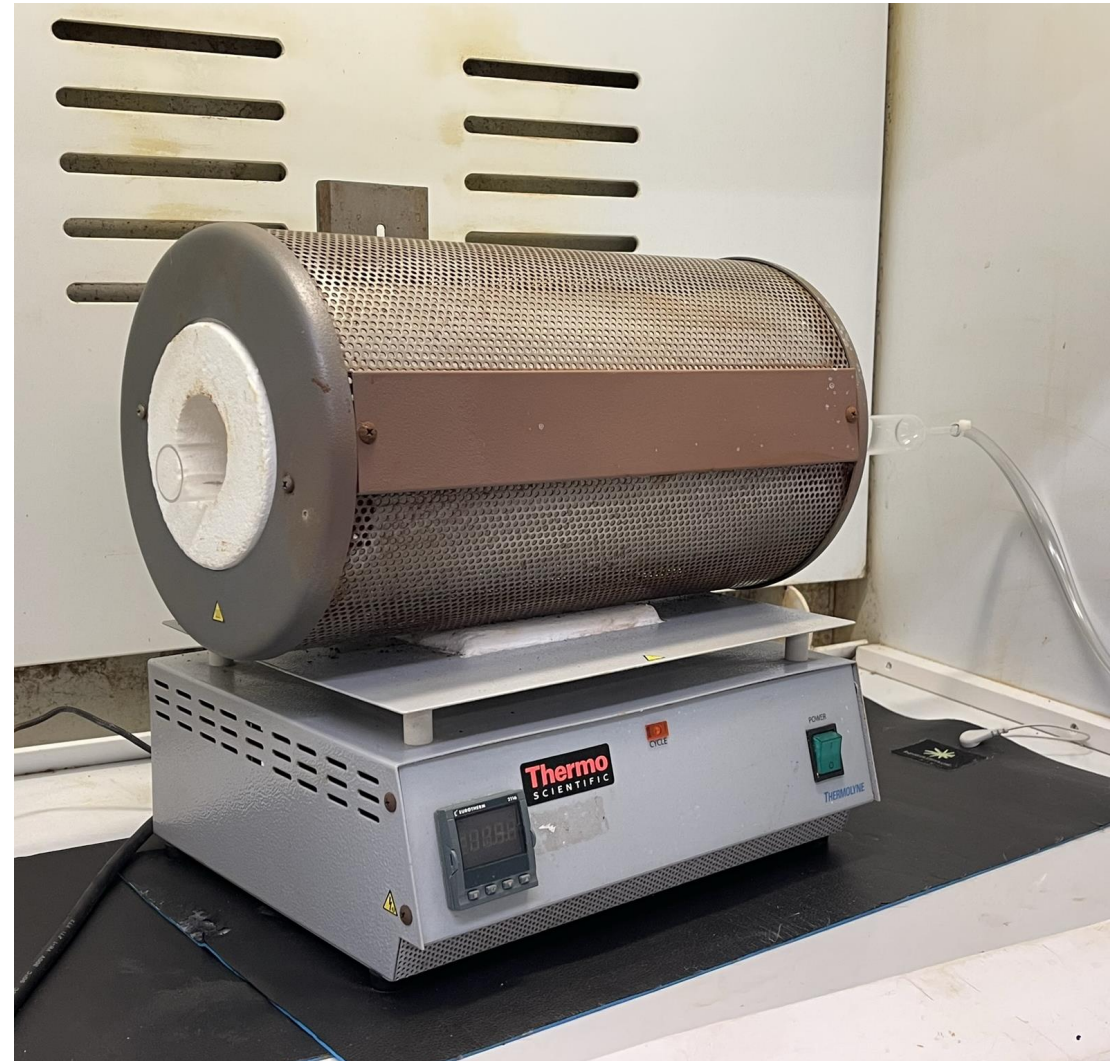


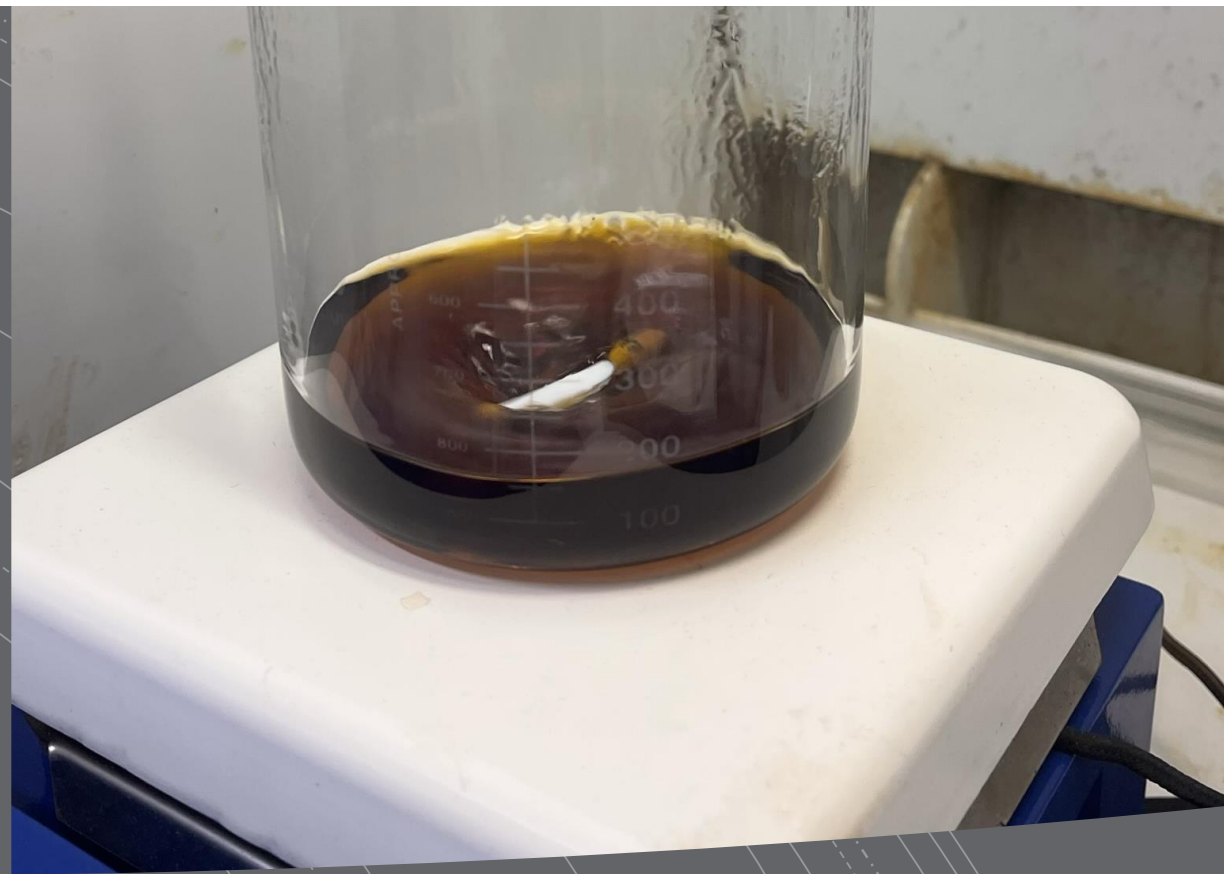
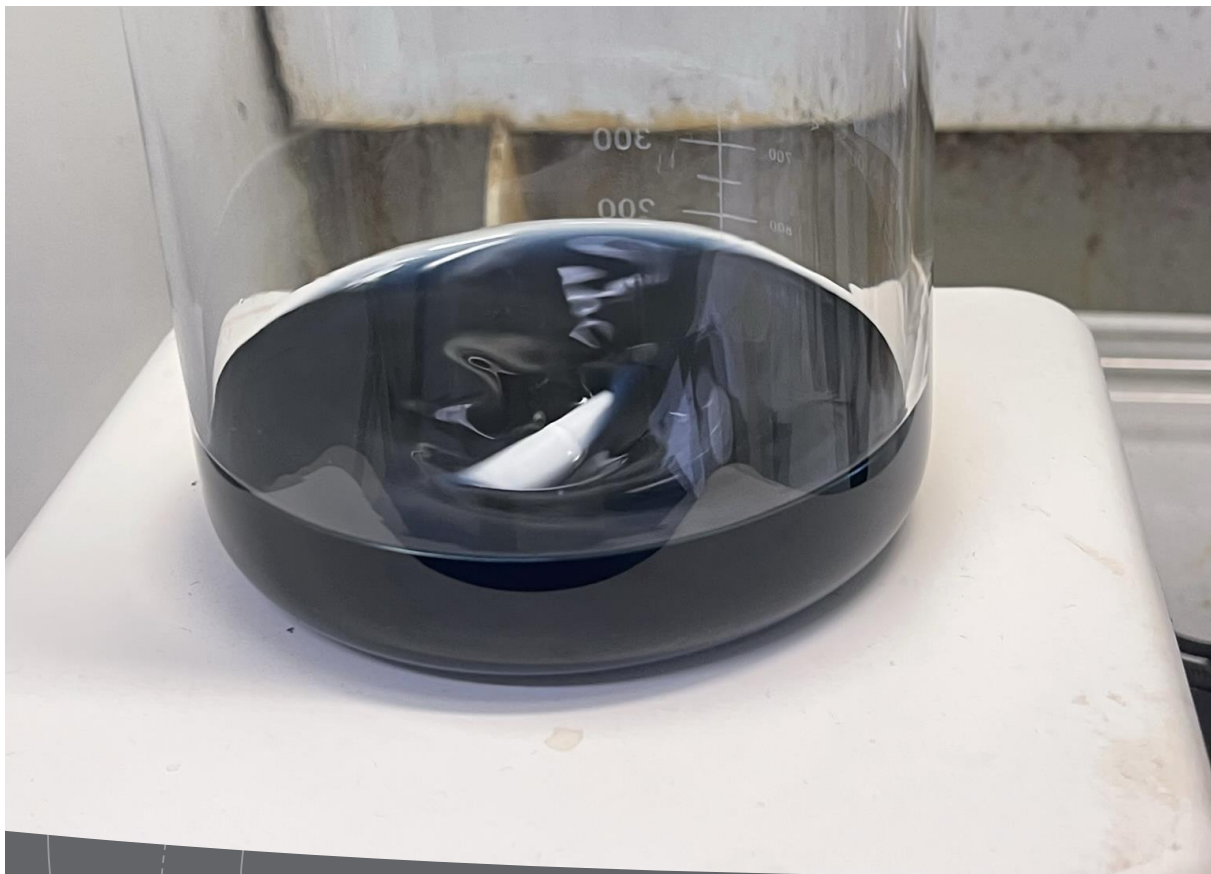
## Major Systematic Errors

- Exposure to air/moisture
- Static

# Gravimetric Reduction

- Reduce the starting material, Os or Ir salts, to Os or Ir metal under  $H_2$  at high temperature.
- Weighing, gravimetry, is the primary measurement procedure.
- SI Traceability: through weighing the salt and metal gravimetric reduction products.
- Determine the mass fraction of metal in the salt.





## Dissolution of the Metal

# NaOH / NaNO<sub>3</sub> Fusion



Sample Os<sup>0</sup> weighed into quartz crucible

NaOH and NaNO<sub>3</sub> added on top of the sample

NaOH and NaNO<sub>3</sub> almost completely melted

NaOH and NaNO<sub>3</sub> completely melted, metal fully dissolved

# Purity Analysis: Requirement for Establishing SI Traceability

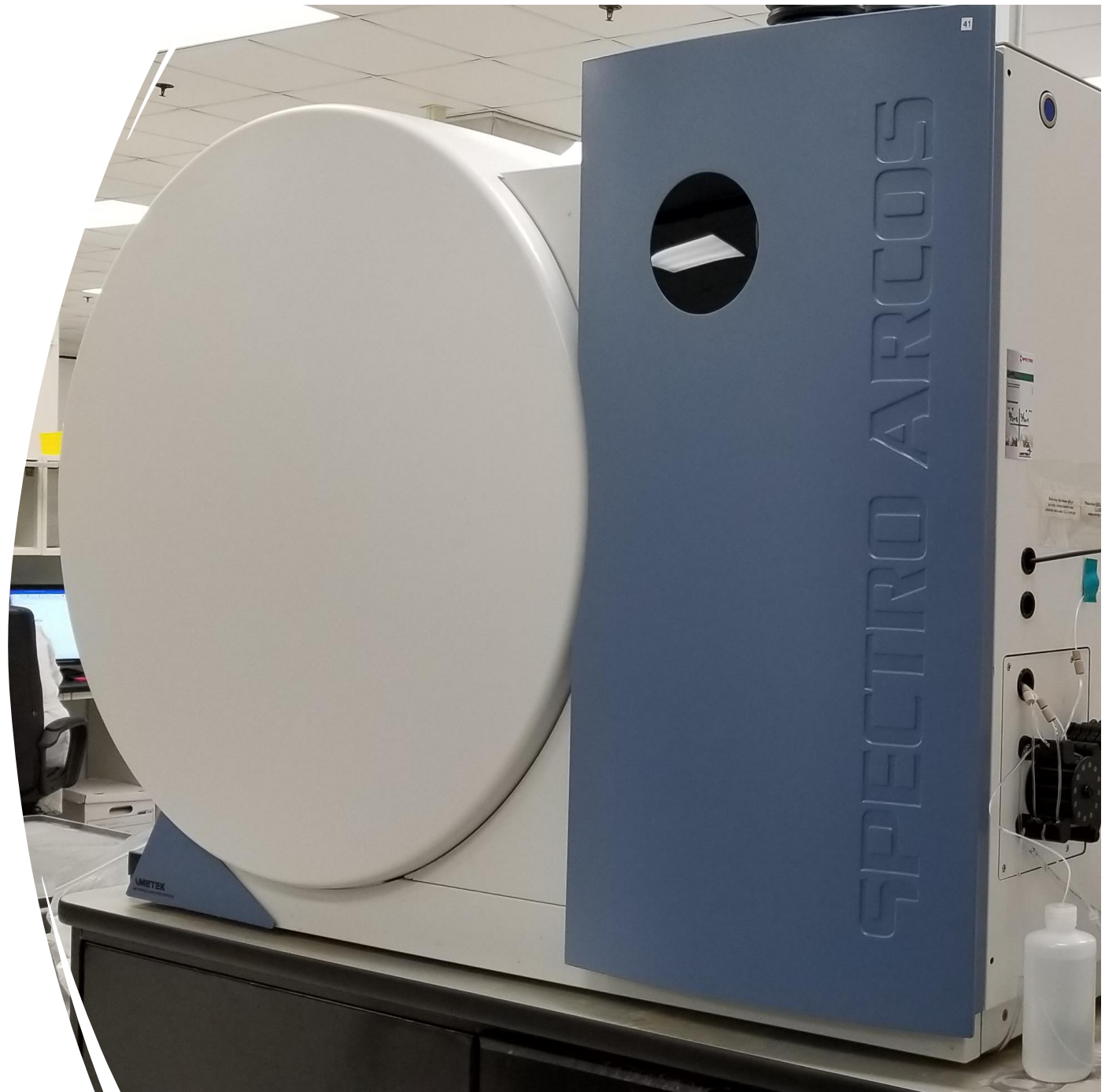
- Trace metallic Impurities (TMI) analysis was performed on solutions of the candidate SI traceable salts
- Inert Gas Fusion analysis was performed to determine estimates of the O, N, and H impurities.
- The combined impurities from TMI and IGF analysis were applied as corrections to the mass fraction of Ir or Os in the starting material

# ICP-OES Comparison

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The metal mass fraction in the salts was found by Gravimetric Reduction.

- Highly reproducible results
- The accuracy of the results must be confirmed.



# ICP-OES Comparison of Solution Standards

Solution standards are made from the two independent starting materials.

- Solution standards are made from the candidate SI traceable salt.
- The comparison solution standard is made from the dissolved high purity Ir or Os metal powders.

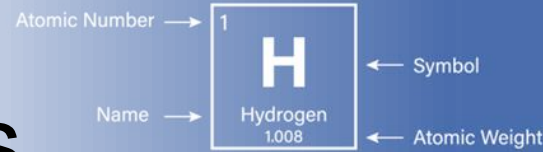




# ICP-OES

## Comparison of Solution Standards

### Periodic Table of the Elements



- **Accurate matrix matching** and/or **internal standardization** of SI traceable solution standard to comparison solution standard
- Careful **monitoring** of metal **washout**
- **Natural aspiration** function of the nebulizer
- Large number of replicates
- Bracketing of samples

|                                    |                                 |                                       |                                     |                                   |                                  |                                    |                                  |                                  |                                    |                                   |                                   |                                     |                                  |                                       |                                   |                                  |                                |
|------------------------------------|---------------------------------|---------------------------------------|-------------------------------------|-----------------------------------|----------------------------------|------------------------------------|----------------------------------|----------------------------------|------------------------------------|-----------------------------------|-----------------------------------|-------------------------------------|----------------------------------|---------------------------------------|-----------------------------------|----------------------------------|--------------------------------|
|                                    |                                 |                                       |                                     |                                   |                                  |                                    |                                  |                                  |                                    |                                   |                                   | 13<br>IIIA                          | 14<br>IVA                        | 15<br>VA                              | 16<br>VIA                         | 17<br>VIIA                       |                                |
|                                    |                                 |                                       |                                     |                                   |                                  |                                    |                                  |                                  |                                    |                                   |                                   | 5<br>B<br>Boron<br>10.81            | 6<br>C<br>Carbon<br>12.011       | 7<br>N<br>Nitrogen<br>14.007          | 8<br>O<br>Oxygen<br>15.999        | 9<br>F<br>Fluorine<br>18.9984031 |                                |
|                                    |                                 |                                       |                                     |                                   |                                  |                                    |                                  |                                  |                                    |                                   |                                   | 13<br>Al<br>Aluminium<br>26.9815385 | 14<br>Si<br>Silicon<br>28.085    | 15<br>P<br>Phosphorus<br>30.973761998 | 16<br>S<br>Sulfur<br>32.06        | 17<br>Cl<br>Chlorine<br>35.45    |                                |
|                                    |                                 |                                       | 22<br>Ti<br>Titanium<br>47.867      | 23<br>V<br>Vanadium<br>50.9415    | 24<br>Cr<br>Chromium<br>51.9961  | 25<br>Mn<br>Manganese<br>54.938044 | 26<br>Fe<br>Iron<br>55.845       | 27<br>Co<br>Cobalt<br>58.933194  | 28<br>Ni<br>Nickel<br>58.6934      | 29<br>Cu<br>Copper<br>63.546      | 30<br>Zn<br>Zinc<br>65.38         | 31<br>Ga<br>Gallium<br>69.723       | 32<br>Ge<br>Germanium<br>72.630  | 33<br>As<br>Arsenic<br>74.921595      | 34<br>Se<br>Selenium<br>78.971    | 35<br>Br<br>Bromine<br>79.904    |                                |
|                                    |                                 |                                       | 38<br>Y<br>Yttrium<br>88.90584      | 40<br>Zr<br>Zirconium<br>91.224   | 41<br>Nb<br>Niobium<br>92.90637  | 42<br>Mo<br>Molybdenum<br>95.95    | 43<br>Tc<br>Technetium<br>(98)   | 44<br>Ru<br>Ruthenium<br>101.07  | 45<br>Rh<br>Rhodium<br>102.90550   | 46<br>Pd<br>Palladium<br>106.42   | 47<br>Ag<br>Silver<br>107.8682    | 48<br>Cd<br>Cadmium<br>112.414      | 49<br>In<br>Indium<br>114.818    | 50<br>Sn<br>Tin<br>118.710            | 51<br>Sb<br>Antimony<br>121.760   | 52<br>Te<br>Tellurium<br>127.60  | 53<br>I<br>Iodine<br>126.90447 |
|                                    |                                 | 68-71<br>Lanthanoids                  | 72<br>Hf<br>Hafnium<br>178.49       | 73<br>Ta<br>Tantalum<br>180.94788 | 74<br>W<br>Tungsten<br>183.84    | 75<br>Re<br>Rhenium<br>186.207     | 76<br>Os<br>Osmium<br>190.23     | 77<br>Ir<br>Iridium<br>192.217   | 78<br>Pt<br>Platinum<br>195.084    | 79<br>Au<br>Gold<br>196.966569    | 80<br>Hg<br>Mercury<br>200.592    | 81<br>Tl<br>Thallium<br>204.38      | 82<br>Pb<br>Lead<br>207.2        | 83<br>Bi<br>Bismuth<br>208.98040      | 84<br>Po<br>Polonium<br>(209)     | 85<br>At<br>Astatine<br>(210)    |                                |
|                                    |                                 | 86-103<br>Actinoids                   | 104<br>Rf<br>Rutherfordium<br>(261) | 105<br>Db<br>Dubnium<br>(268)     | 106<br>Sg<br>Seaborgium<br>(269) | 107<br>Bh<br>Bohrium<br>(270)      | 108<br>Hs<br>Hassium<br>(269)    | 109<br>Mt<br>Meitnerium<br>(278) | 110<br>Ds<br>Darmstadtium<br>(281) | 111<br>Rg<br>Roentgenium<br>(282) | 112<br>Cn<br>Copernicium<br>(285) | 113<br>Nh<br>Nihonium<br>(286)      | 114<br>Fl<br>Flerovium<br>(289)  | 115<br>Mc<br>Moscovium<br>(289)       | 116<br>Lv<br>Livermorium<br>(293) | 117<br>Ts<br>Tennessine<br>(294) |                                |
| 57<br>La<br>Lanthanum<br>138.90547 | 58<br>Ce<br>Cerium<br>140.116   | 59<br>Pr<br>Praseodymium<br>140.90766 | 60<br>Nd<br>Neodymium<br>144.242    | 61<br>Pm<br>Promethium<br>(145)   | 62<br>Sm<br>Samarium<br>150.36   | 63<br>Eu<br>Europium<br>151.964    | 64<br>Gd<br>Gadolinium<br>157.25 | 65<br>Tb<br>Terbium<br>158.92535 | 66<br>Dy<br>Dysprosium<br>162.500  | 67<br>Ho<br>Holmium<br>164.93033  | 68<br>Er<br>Erbium<br>167.259     | 69<br>Tm<br>Thulium<br>168.93422    | 70<br>Yb<br>Ytterbium<br>173.045 | 71<br>Lu<br>Lutetium<br>174.9668      |                                   |                                  |                                |
| 89<br>Ac<br>Actinium<br>(227)      | 90<br>Th<br>Thorium<br>232.0377 | 91<br>Pa<br>Protactinium<br>231.03688 | 92<br>U<br>Uranium<br>238.02891     | 93<br>Np<br>Neptunium<br>(237)    | 94<br>Pu<br>Plutonium<br>(244)   | 95<br>Am<br>Americium<br>(243)     | 96<br>Cm<br>Curium<br>(247)      | 97<br>Bk<br>Berkelium<br>(247)   | 98<br>Cf<br>Californium<br>(251)   | 99<br>Es<br>Einsteinium<br>(252)  | 100<br>Fm<br>Fermium<br>(257)     | 101<br>Md<br>Mendelevium<br>(258)   | 102<br>No<br>Nobelium<br>(259)   | 103<br>Lr<br>Lawrencium<br>(266)      |                                   |                                  |                                |



# ICP-OES Comparison Results

- Results for candidate SI solution standards were  $< 0.1\%$  from the calculated values.
- Demonstrated the accuracy of the candidate SI traceable solution standards.
- The high precision of the analysis lowered the ICP-OES uncertainty.

A hand is shown pointing at a tablet screen. The screen displays a bar chart with several vertical bars of varying heights. The entire image has a blue overlay. The title 'Uncertainty Estimates' is positioned in the upper right area of the image.

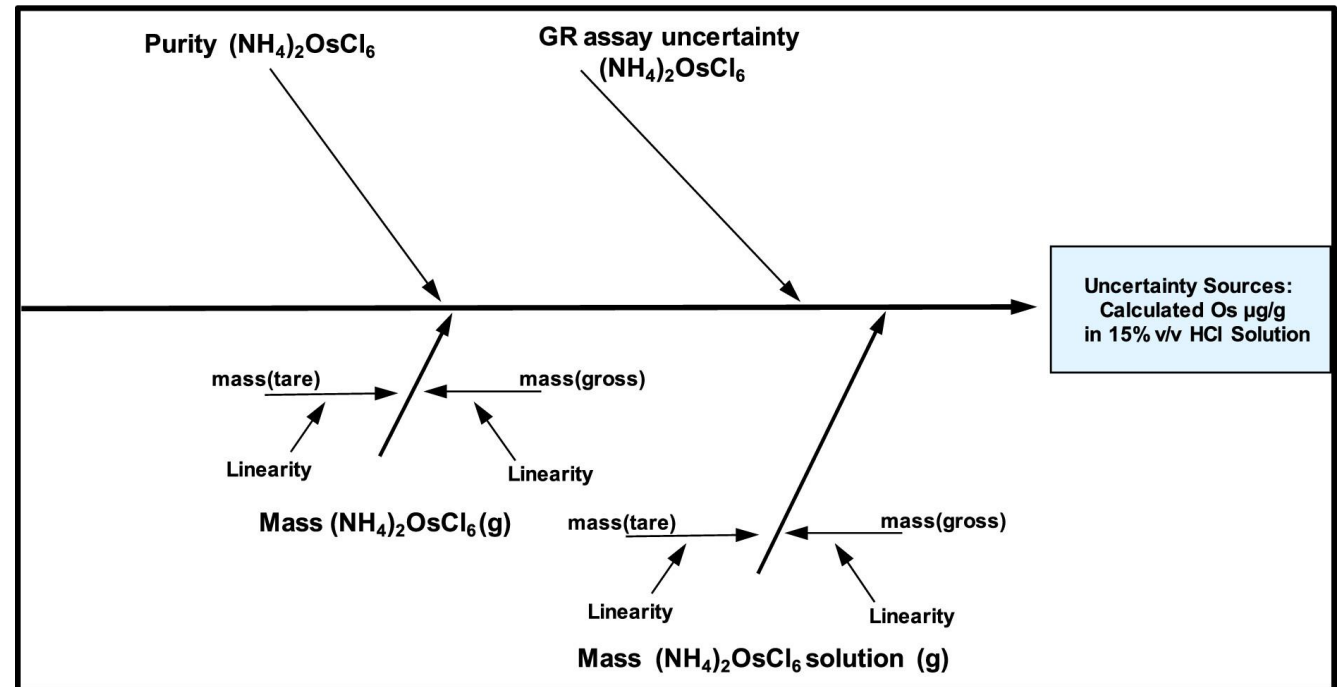
# Uncertainty Estimates

- ISO 17034 and ISO 17025 accreditation requires reporting the uncertainty in a certified value
- Make uncertainty estimates using:
  - Cause and Effect diagrams
  - Error budget analysis.

# Uncertainty Estimates

## Cause and Effect diagram

Uncertainty sources in the preparation of the Os Solution standard in 15% v/v HCl.



# Uncertainty Estimates

## Error Budget Analysis

|   | Error Budget Contributions:<br>Preparation of Os Solution Standard<br>in 15% v/v HCl | Individual Uncertainty Contributions<br>( $u_x$ ) | Value (X) | Relative Uncertainty<br>$= (u_x / X)^2$ |
|---|--|---|-----------|---|
| GR assay uncertainty<br>(NH <sub>4</sub> ) <sub>2</sub> OsCl <sub>6</sub> | Precision of GR assay  |   |           | 1.27E-06                                |
| Purity<br>(NH <sub>4</sub> ) <sub>2</sub> OsCl <sub>6</sub>               | Purity   | 0.0002  | 0.9979    | 4.02E-08                                |
| Mass<br>(NH <sub>4</sub> ) <sub>2</sub> OsCl <sub>6</sub> (g)             | Weight of<br>(NH <sub>4</sub> ) <sub>2</sub> OsCl <sub>6</sub>                       | 0.0001g   | 1.16g     | 4.95E-09                                |
| Mass<br>(NH <sub>4</sub> ) <sub>2</sub> OsCl <sub>6</sub> solution (g)    | Weight of<br>(NH <sub>4</sub> ) <sub>2</sub> OsCl <sub>6</sub> solution              | 0.02g   | 500g      | 1.07E-09                                |
|   | CRU = Combined Relative Uncertainty<br>$= (\sum (u_x/x)^2)^{0.5}$                    |   |           | 1.15E-03                                |
| Expanded Uncertainty of Characterization                                  | Expanded Uncertainty = 2 * Os (μg/g) * CRU =<br>2*1000*0.00115                       |   |           | 2                                       |
|   | Os (μg/g)<br>Calculated  |   |           | 1000                                    |

Error Budget for the Preparation of the Os Solution Standard in 15% v/v HCl

# In Summary



Primary SI traceable solution standards  
for Os and Ir

Certified Value ( $\mu\text{g/g}$ )  
 $\pm$  Expanded Uncertainty ( $\mu\text{g/g}$ )

Low uncertainties:  
fit for purpose as  
Calibration Standards

ISO 17034 and ISO 17025 accreditation